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required: 560.4077. ¹H NMR (C_5D_5N): δ 0.59 (3H, s), 0.73 (3H, s), 0.88 (3H, d, J = 6.6 Hz), 0.89 (3H, d, J = 6.6 Hz), 0.99 (3H, d, J = 6.8 Hz), 1.08 (3H, d, J = 6.6 Hz), 4.01 (1H, m, 3-H), 4.05–4.32 (4H, m, 2'-H, 3'-H, 4'-H, and 5'-H), 4.39 (1H, dd, J = 11.5 and 5.3 Hz, 6'-H), 4.60 (1, dd, J = 11.5 and 2.4 Hz, 6'-H), 5.30 (3H, m, 7-H, 22-H, and 23-H). ¹³C NMR (C_5D_5N): Table 1.

Compound 3. Colourless plates, $[\alpha]_D^{2^2} + 19.8^\circ$ (EtOH; c 0.2), IR $v_{\text{max}}^{\text{RBr}}$ cm⁻¹: 3360 (OH), 1640 and 1530 (CONH), 1460, 1080. EIMS m/z (rel. int.): 276 $[C_{18}H_{46}N]^+$ (57), 258 (28), 180 $[g]u]^+$ (10). FABMS m/z (rel. int.): 778 $[M_{II} + Na]^+$ (18), 750 $[M_I + Na]^+$ (25), 576 $[M_{II} - g]u + H]^+$ (12), 548 $[M_I - g]u + H]^+$ (26), 294 $[C_{18}H_{49}NO]^+$ (10), 276 $[C_{18}H_{47}N]^+$ (13). ¹H NMR (C_5D_5N) : δ 0.89 (3H × 2, br t), 1.29 (ca 42H), 1.63 (3H, s), 2.04 (2H, br t), 2.19 (4H, br t), 3.90 (1H, m), 4.04 (1H, br t), 4.1–4.8 (9H, m), 4.94 (1H, d, J = 7.6 Hz, anomeric H), 5.28 (1H, m), 6.00 (2H, br t), 6.02 (2H, m), 8.39 (1H, d, J = 7.8 Hz, CONH). ¹³C NMR (C_5D_5N) : δ 14.0 × 2 (q), 16.0 (q), 22.7 × 2 (t), 25.7 (t), 28.0 (t), 28.2 (t), 29.5 × 3 (t), 30.1 × 10 (t), 32.1 × 2 (t), 33.0 (t), 35.5 (t), 39.9 (t), 54.4 (d), 62.4 (t), 69.5 (t), 71.2 (d), 72.0 (d), 72.3 (d), 74.6 (d), 77.9 × 2

(d), 105.6 (d), 124.0 (d), 131.5 (d), 132.2 (d), 135.5 (s), 175.5 (s).

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p-HYDROXY ACETOPHENONE DERIVATIVES FROM DIOSCOREA BULBIFERA

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Key Word Index—Dioscorea bulbifera; Dioscoreaceae; bulbs; p-hydroxy acetophenone derivatives.

Abstract—From the bulbs of *Dioscorea bulbifera*, two new *p*-hydroxy acetophenone derivatives, namely 4-hydroxy-[2-trans-3',7'-dimethyl-octa-2',6'-dienyl]-6-methoxyacetophenone and 4,6-dihydroxy-2-O-(4'-hydroxybutyl)acetophenone have been isolated.

INTRODUCTION

The genus Dioscorea comprises 600 tropical and subtropical species; three species are distributed in the Pyrenees, Balkan Penins and Caucasus [1]. Dioscorea bulbifera is common throughout India ascending up to 6000 ft in the Himalayas. Its bulbs are used to treat piles, dysentry, syphilis and are applied to ulcers [2]. Poisonous glucosides have already been reported from bulbs of Dioscorea [3]. This paper describes the isolation of phydroxy acetophenone derivatives from D. bulbifera; their structures were established by spectroscopic methods.

RESULTS AND DISCUSSION

A series of conventional extraction and separation procedures yielded compounds 1 and 2. Homogeneity and purity of these compounds was established by chromatography.

Compound 1 shows a $[M]^+$ in the mass spectrum at m/z 302 in agreement with the formula $C_{19}H_{26}O_3$. Its UV spectrum is characteristic of acetophenone derivatives. Acetylation of 1 yielded a monoacetate (1a), showing the presence of one hydroxyl group in the molecule. The bathochromic shift (10 nm) induced in the UV spectrum

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$$\begin{array}{ccc} \mathbf{I} & \mathbf{R} = \mathbf{H} \\ \mathbf{Ia} & \mathbf{R} = \mathbf{A}c \end{array}$$

2 R = H 2a R = Ac

of 1 on addition of sodium acetate to an ethanolic solution and no shift with aluminium chloride, is indicative of the presence of an -OH group para to a -COMe group [4, 5].

Compound 1 possessed one methoxy group [¹H NMR δ 3.59 (s, 3H)] and one 3':7'-dimethyl-octa-2':6' dienyl group [¹H NMR δ 1.60 (3H, s), 1.65 (3H, s), 1.79 (3H, s), 2.05 (4H br m), 3.30 (2H, d, J = 6.2 Hz), 5.10-5.15 (1H, br t, J = 7 Hz), 5.17-5.28 (1H, br q, J = 1.0 and 7.1 Hz). The multiplicity and J values of the 2' vinylic proton suggested a trans configuration for the 2' double bond. The ¹H NMR spectrum also gave two signals in the aromatic region, [δ 5.70 (d, 1H, J = 2.5 Hz) and 5.85 (d, 1H, J = 2.4 Hz)]. These two protons are meta coupled as indicated by the spectrum. The compound, therefore, is p-hydroxy acetophenone having methoxy and 3',7'-dimethyloctadienyl groups as substituents. These two substituents either must be present at positions C-2 and C-6 or C-3 and C-5.

The protons at C-2 and C-6 are deshielded (*ortho* to carbonyl group), so they should resonate at higher δ value (δ 7.75 and 7.73 ppm) [7], and protons at C-3 and C-5 are shielded so they should resonate at lower δ value (δ 5.68 and 5.82 ppm) [8]. Since the protons in the ¹H NMR of 1 occur at δ 5.70 and 5.85 the substituents must be present at C-2 and C-6. The position of substituents at the other positions was further confirmed by the ¹H NMR data of the corresponding acetate (1a), which suggested that the two aromatic protons must be present at C-3 and C-5 because such protons showed a considerable downfield shift (from δ 5.70–6.02 and 5.85–6.20) upon acetylation of a phenolic hydroxyl group [4].

Compound 1 was thus established to be 4-hydroxy-2-[3',7'-dimethyl-octa-2',6'-dienyl]-6-methoxyacetophenone. This structure was further supported by its mass spectrum and ¹³C NMR spectrum.

Compound 2 has a $[M]^{\frac{1}{4}}$ at m/z 240 in agreement with the formula $C_{12}H_{16}O_5$. Its UV spectrum was similar to

that of 1. The shift induced in its UV spectrum on addition of sodium acetate [12 nm], as well as on addition of AlCl₃/HCl (20 nm), indicated the presence of a free hydroxyl group at *ortho* and *para* position with respect to the -COMe group [4]. Acetylation of 2 yielded a triacetate (2a) showing the presence of three hydroxyl groups.

The ¹H NMR of 2 was consistent with a tetrasubstituted aromatic ring with an acetate group, one Ohydroxybutyl residue and two phenolic hydroxyl groups as substituents. The structure of the hydroxybutyl side chain was established unambiguously by ¹H NMR. There are two -OCH₂ groups in the side chain. One shows up as a triplet of 2H and the other is represented by two diastereotopic protons with different chemical shifts. The latter must be associated with a -CH2OH group, since in very pure NMR, a direct coupling of the two diastereomeric oxymethylene protons with the hydroxy protons is observed [9]. The other -OCH₂ is therefore directly linked to the acetophenone ring by an aromatic linkage. The broad triplets $-O-CH_2-CH_2-CH_2-CH_2OH$ were observed at δ 1.79 (2H, br t), and (1.59, 2H br t). Again, the appearance of meta coupled aromatic protons as doublets at lower δ value (δ 5.65 and 5.80) ppm indicated that the compound is a 2, 4, 6-trisubstituted acetophenone [8]. Compound 2 was thus established to be 4,6-dihydroxy-2-O-(4'hydroxybutyl)acetophenone; its structure was further supported by ¹³C NMR.

EXPERIMENTAL

Mps: uncorr. Analytical TLC was carried out on silica gel G [Merck 7731] with (i) EtOAc, (ii) MeOH-CHCl₃ (1:1) unless otherwise stated. CC was done on silica gel 60 [Merck 7734]. UV spectra were recorded in EtOH, IR as KBr disks. ¹H NMR spectra were measured at 90 MHz in CDCl₃ soln, unless otherwise specified, using TMS as int. std. ¹³C NMR were recorded at 25.05 MHz in C₅D₅N soln.

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		3,0,711		
Н	1	1a	2	2a
3	5.70 d (2.4 Hz)	6.02 d (2.5 Hz)	5.65 d (2.5 Hz)	6.00 d (2.5 Hz)
5	5.85 d (2.4 Hz)	6.20 d (2.5 Hz)	5.80 d (2.5 Hz)	6.31 d (2.5 Hz)
8	2.45 s	2.40 s	2.48 s	2.35 s
1'	3.30 d (6.2 Hz)	3.31 d (6.2 Hz)	4.30 t	4.31 t
2'	5.17-5.28 (br q, J=7.1, 1.0 Hz)	5.17-5.28 (br q, J = 1.0, 7.1 Hz)	1.79 t	1.80 t
3′			1.59 t	1.62 t
4′	2.05~(br~m)	$2.06 \ (br \ m)$	$H_A 4' 3.65 dd$ (11.25, 3.40 Hz)	4.29 dd (4.79, 4.45)
			H _B 4' 3.58 dd (11.25, 7.35 Hz)	4.19 dd (11.80, 5.80)
5'	2.05 (br m)	$2.06 \ (br \ m)$	_	
6'	5.10-5.15 (br t, $J = 7.0$ Hz)	5.10-5.15 (br t) J = 7.0 Hz	_	_
8'	1.60 s	1.62 s	_	
9′	1.65 s	1.65 s	_	
10′	1.79 s	1.80 s	_	_
Ph-OH	7.50		7.90	
Ph-OH	_	·	13.10	_
-CH ₂ OH			2.62 br t (6.2 Hz)	_
Ph-OMe	3.59 s	3.61 s	_	
Ph-OAc	_	2.16 s	-	2.11 s
Ph-OAc				2.21 s
-CH ₂ OAc		-	-	2.02 s

Table 1. ¹H NMR of compounds 1, 1a, 2 and 2a (90 MHz, CDCl₃) (δ , ppm)

Plant material. D. bulbifera bulbs were collected in Nainital, India in January 1986 (a herbarium specimen of the plant is on file at the Botanical Survey of India, Allahabad).

Extraction and separation. Dried and ground bulbs (1 kg) were soxhlet extracted with EtOH. Extracts was sepd into H₂O sol. and H₂O insol. fractions. The H₂O insol. portion was sepd by flash CC and then eluted with different solvents of increasing polarity. Elution with MeOH-CHCl₃ (1:4) yielded 1 (900 mg) and elution with MeOH-CHCl₃ (2:3) yielded 2 (40 mg).

Compound 1. Mp 225°, homogenous on TLC, R_f 0.71 (solvent i), 0.72 (solvent ii). Found C 75.49%, H 8.60%, calculated for $C_{19}H_{26}O_3$, C 75.5%, H 8.62%. UV λ_{max}^{E1OH} nm: 220, 285; +AlCl₃/HCl: 240: 287; +NaOAc: 230, 289. IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3200, 2900, 2825, 1660, 1600, 1530, 1450, 1380, 1137, 1105, 880, 825, 780. ¹H NMR [90 MHz, CDCl₃] see Table 1. ¹³C NMR; δ_c 160.1 (s, C-4), 120.2 (s, C-1), 162.40 (s, C-2), 167.90 (s, C-6), 113.23 (d, C-5), 116.31 (d, C-3), 184.70 (s, Me-CO-), 28.90 (q, Me-CO), 29.01 (t, C-1'), 128.52 (d, C-2'), 136.10 (s, C-3'), 41.10 (t, C-4'), 27.90 (t, C-5'), 127.22 (d, C-6'), 135.70 (s, C-7'), 17.52 (q, C-8'), 17.10 (q, C-9'), 16.40 (q, C-10'), 56.29 (O-Me). EIMS, m/z (rel. int. %): 302 $[M]^+$, 287 $[M-Me]^+$ (5), 271 $[M-31]^+$ (100), 233 $[M]^+$ $-C_5H_9$]⁺ (40), 210 (10), 165 [M- $C_{10}H_{17}$]⁺ (90), 150 [165] -Me] (50), 104 (35), 91 (15), 76 (25). Acetylation of 1 (40 mg) yielded 1a, mp 213°. Found: C73%, H 8%, calculated for $C_{21}H_{28}O_4$, C 73.25%, H 8.13%. IR v_{max}^{KBr} cm⁻¹: 1760, 1260 (acetate), 1655 (C=O), 1600, 1520 (aromatic), 1180, 1010, 880, 820 cm $^{-1}$. MS m/z: 344 [M] $^{+}$. 1 H NMR see Table 1.

Compound 2. Mp 280°, homogeneous on TLC, $R_f = 0.25$ [solvent (ii)]. Found: C 60%, H 6.7%, calculated for $C_{12}H_{16}O_5$, C 60%, H 6.66%. UV $\lambda_{max}^{\text{EiOH}}$: 230, 290; +AlCl₃/HCl 240, 287;

+ NaOAc 242, 287. IR $v_{\rm max}^{\rm KBr}$ cm $^{-1}$: 3380, 3200, 2918, 2850, 1650, 1540, 1450, 885, 825, 780. 1 H NMR see Table 1. MS m/z 240 [M] $^{+}$. 13 C NMR: $\delta_{\rm c}$ 161.0 (s, C-4), 119.2 (s, C-1), 163.10 (C-2), 162.50 (s, C-6), 114.51 (d, C-5), 116.90 (d, C-3), 182.90 (s, COMe), 29.0 (q, COMe), 72.9 (t, OCH₂), 69.1 (t, OCH₂), 34.2 (t, C-2'), 32.9 (t, C-3'). Acetylation of 2 yielded 2a. Found: C 59.2%, H 6.01%, calculated for C₁₈H₂₂O₈ C 59%, H 6%. IR $v_{\rm max}^{\rm KBr}$ cm $^{-1}$: 1760, 1680, 1600, 1580, 1370, 880, 810. 1 H NMR see Table 1. MS m/z 366 [M] $^{+}$.

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